THEKNOWLEDGE Year 7 B1.3 Reproduction

Adolescence: During adolescence your body goes through physical changes; this is called **puberty**.

Changes that happen to both boys and girls include:

• Pubic hair and under arm hair grows, body smell becomes stronger, you experience emotional changes and have a growth spurt.

Changes in girls:		Changes in boys:		
•	Breast develop, hips widen, period start. Ovaries start releasing egg cells	 Voice breaks Testes and penis get bigger Testes start making sperm Hair grows on face and chest 		

Sentence starters:

- During sexual intercourse the
- Substances are passed between a mother and her fetus



Reproductive Systems:



The Menstrual Cycle:

Day 1: The uterus lining breaks down. Day 5 : Bleeding stops, a **period**. The uterus lining starts to re-grow Day 14: An egg cell is released from one of the ovaries, this is **ovulation**.



Flowers, Pollination, Fertilisation, Germination and Seed Dispersal:



Seeds can be
dispersed by: animals,
water, wind or
explosive dispersal.
Seeds need water,
warmth and oxygen for
germination.



of the pollen The pollen nucleus moves down the tube. The pollen nucleus joins with the ovule nucleus. Fertilisation takes place and a seed will form.

THEKNOWLEDGE Year 7 C1.3 Reactions

Chemical reactions:

A chemical reaction is a change in which atoms are rearranged to create new substances.

Most chemical reactions are not easily **reversible**. At the end of the reaction it is very difficult to get back what you started with. How do you know if a chemical reaction has happened?

- 1) See huge flames or tiny sparks
- 2) Notice a sweet smell or foul stink
- 3) Feel the chemicals getting hotter or colder
- 4) Hear a loud bang or gentle fizzing

Some reactions happen quickly, others are much slower. Chemists use **catalysts** to speed up slow reactions. A catalyst is not used up in a reaction.

<u>Word equations</u>: In a chemical reaction, the starting substances are called **reactants** and the substances that are made are called **products**.

A word equation shows the reactants on the left and the products on the right.

The arrow means '*reacts to make*'.

The word equation for the reactions of iron and sulphur is:

iron + sulfur → iron sulphide

Rearranging atoms:

When magnesium burns in air, it reacts with oxygen to form magnesium oxide.

The reactants and products look different.

Reactants – shiny magnesium and invisible oxygen gas Products – white magnesium oxide powder

In every chemical reaction, the atoms get rearranged.



Sentence starters:

A chemical reaction is when.... We test for carbon dioxide by.... In any chemical reaction the total mass of reactants is...

<u>Glossary</u>

Reversible Reactants Products Combustion Catalyst Fossil fuels Non-renewable Thermal decomposition Conservation Endothermic Exothermic

Burning fuels:

A **fuel** is a substance that burns to transfer energy by heating. Burning is also called **combustion**. When something undergoes combustion there are usually two products:

methane + oxygen → carbon dioxide + water

Petrol, diesel and coal are **fossil fuels**, they a **non-renewable** meaning they cannot be replaced once they are used up.

Oxidation, is when substances react with oxygen. Rusting is an example of oxidation, where iron reacts with oxygen and water.

Thermal decomposition:

A **decomposition** reaction is when a compound breaks down into simpler compounds or elements A **thermal decomposition** reaction is when the substance breaks down when it is heated.

Copper carbonate is a green compound, when you heat it, it breaks down into copper oxide and carbon dioxide.

copper carbonate → copper oxide + carbon dioxid

We can prove that the gas is carbon dioxide by bubbling it through limewater and the limewater will go cloudy.



Conservation of mass:

In any chemical reaction the total mass of the reactants is equal to the total mass of products.

Endothermic and exothermic:

Endothermic – the surroundings transfer energy t substances e.g. melting and boiling. Exothermic – energy is transferred to the surroundings e.g. burning wax.

Space Year 7 P1 Space

The Night Sky

We can see artificial satellites using the naked eye. The Moon is the Earth's natural satellite, it orbits the Earth. Meteors are bits of dust or rock that burn up as they move through the Earth's atmosphere. Any meteor that makes it to earth is known as a meteorite.



Lots of the dots we see in the sky are stars in our galaxy, the Milky Way.

The Solar System

The solar system is made up of 4 inner planets and 4 outer planets. All the planets orbit the Sun. The inner planets are all terrestrial planets; they are made of rock. The outer planets are gas giants. Many planets have moons which orbit them. Saturn has 60 moons but Earth has only one.



There are thousands of pieces of rock in the asteroid belt, some are tiny specs of dust but one is large enough to be called a dwarf planet. Ceres is the only dwarf planet inside the orbit of Neptune.

Sentence starters: A satellite is an object that orbits..... The solar system is made up of ... Day and night is caused by... The phases of the moon are....

The Moon



Half the moon is lit up by the Sun all the time. As the Moon moves round the Earth

The moon takes 27 days and 7 hours to orbit the Earth once.

it looks different from the Earth. The changes in shape of the Moon as known as the phases of the Moon.

The phases of the moon include new, crescent, first quarter, gibbous, full, gibbous, third quarter and crescent.

A lunar month is the time from one new moon to the next.

A solar eclipse happens when the Moon blocks the light from the Sun.

A lunar eclipse happens when the Earth comes between the Sun and the Moon.

The Earth

There is day and night on Earth because Earth spins on its axis. It take 24 hours to complete one full spin. The Sun rises in the East each morning and sets in the West. The Earth moves round the Sun once a year.

The Earth's axis is tilted to 23.4°. The days are longer in the Summer and the sun is higher in the sky.



Glossary Satellites Planets The sun Orbit Eclipse The moon Constellations Lunar eclipse Solar eclipse Stars

THEKNOWLEDGE Light	Year 7 P1.3	Sentence starters: The sun is luminous because The difference between specular and o A prism disperses white light to	diffuse reflection is
Light travels in straight lines and from them. Light sources are lur objects like glass transmit light. some light through them but yo objects allow no light to pass the Light travels very quickly. It mov about a million times faster than medium to carry it so it can trav	we can see objects because light reflects ninous and they emit light. Transparent Translucent objects like frosted glass allow u cannot make out any details. Opaque rough them. es at a speed of 300 000 000 m/s. This is a sound. Unlike sound, light does not need a el through a vacuum.	Glossary Angle of incidence Angle of reflection Convex Dispersion Incident ray Inverted Law of reflection Luminous Medium Opaque Prism Refraction Specular reflection transmit	<u>Colour:</u> The colours of the rainbow are all contained in white light.White lightRed Glass PrismRed, blue and green are called the primary colours. Mixing them gives us secondary
Reflection of Light: incidence	reflected ray	The Eye and the Camera:	Colour filters:
The angle of incidence is equal to reflection. Reflection from a smooth Reflection from a rough surface Refraction of Light: Light changes direction when it called refraction.	o the angle of reflection. This is the law of both surface is called Specular Reflection. is called Diffuse Scattering. passes from one medium to another. This is	A How an image is formed in your eye. Our eyes focus light onto the retina at the back of the eye. This is an inverted image. Only light travelling in the right direction gets through the pinhole.	These transmit one colour and absorb the rest.
entering air glass ray emerging	racted	 An image is formed in a pinhole camera. Pinhole Cameras form an image in exactly the same way. Modern digital cameras record images on a CCD 	What makes objects different colours? They reflect one colour while they absorb all the other colours.
converge at a focus or focal point	nt.	(Charge Couple Device).	other colours.